

PHYTOREMEDIATION – A Novel Technology to Decontaminate Polluted Sites



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Phytoremediation

Phytoremediation is defined as the use of green plants and their associated microorganisms, soil amendments, and agronomic techniques to remove, degrade or detoxify harmful environmental pollutants.

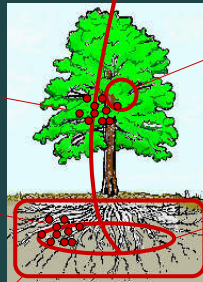
Phytoremediation technologies:

- I. Rhizosphere Enhanced Bioremediation (or Phytostimulation)
- II. Phytodegradation (or Phytotransformation)
- III. Phytostabilization
- IV. Phytoextraction (or Phytoaccumulation)
- V. Rhizofiltration
- VI. Phytovolatilization
- VII. **Phytoexcretion (?)**

Phytoremediation processes

Phytoextraction: transfer of pollutants from the soil and accumulation in the above ground parts of the plant.

Rhizofiltration: transfer of pollutants from the soil and accumulation in the roots of the plant.



Phytovolatilization : transfer of pollutants from the soil to the atmosphere.

Phytodegradation: enzymatic degradation of the pollutants in the plant tissue.

Phytostabilization: Stabilization of heavy metals in the soil/root surface and reduction of heavy metal mobility.

Enhanced Bioremediation (or Phytostimulation) : Enhancement of the microbial community and increase of biodegradation in the rhizosphere.

Phytoremediation Research at TU-Crete

General Project:

Phytoremediation of contaminated sites with heavy metals using Mediterranean plants.

Specific aims:

- Heavy metals: Lead (Pb), Cadmium (Cd) and their mixtures.
- Identification of Pb and Cd hyperaccumulators among Mediterranean plants
- Focusing on salt-tolerant plants

Why halophytes??

- Halophytes can be **cultivated with saline irrigation water** which is a desirable feature since often high-quality irrigation water is not available even for application to crops in arid and semi-arid regions.
- **Salt-water irrigation** is becoming an increasingly important practice because the quality of irrigation waters is decreasing as water supplies for agriculture become restricted due to urban needs and climate change.
- **Salinity** has been shown to be a **key factor** for
 - the increased bioavailability of metals in the soils due to reduced soil metal sorption
 - the translocation of metals from roots to the aerial parts of the plant - an important feature for phytoextraction applications

Salt-tolerant plants examined:

Plant #1: *Tamarix smyrnensis*



Plant #2: *Nerium oleander*

Plant #3: *Atriplex halimus*



Tamarix – Experiments

Pot experiments with plants grown in metal polluted soils in order to evaluate the effect of metals and soil salinity on the growth of plant

Measurements:

Plant

- ➔ Biomass
- ➔ Height
- ➔ Water content
- ➔ Chlorophyll
- ➔ Proteins
- ➔ Peroxidase activity
- ➔ Metal content (in roots and shoots)

Soil

- ➔ Total metals
- ➔ Plant available metals
- ➔ pH
- ➔ EC
- ➔ Organic matter
- ➔ Total CaCO₃

Pot Experiments

Experimental Conditions

T. smyrnensis growing in contaminated soil with 800 ppm Pb and 16 ppm Cd

10 -15 cm cuttings of *T. smyrnensis*

Propagation period : 21 days

Adaptation period : 8 months

Experimental period : 10 weeks

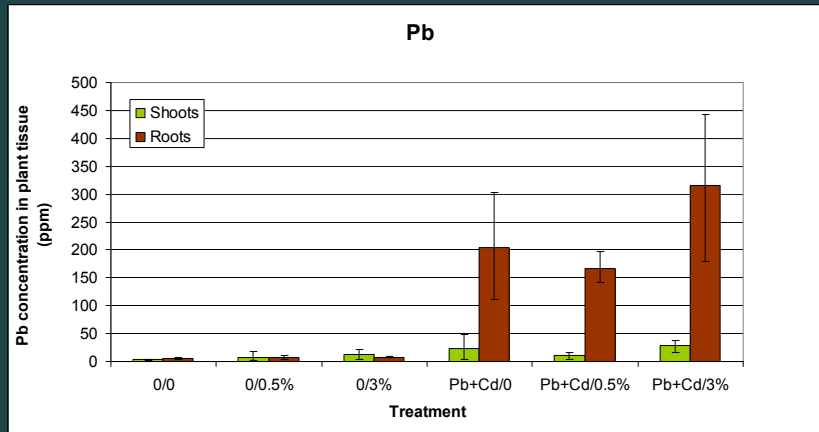
Temperature : 19 – 47°C

Humidity : 18 – 70%

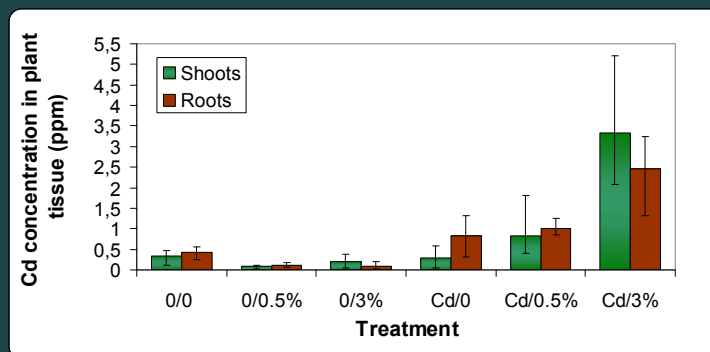
Photoperiod : 14 -15 h



Pb accumulation in the plant



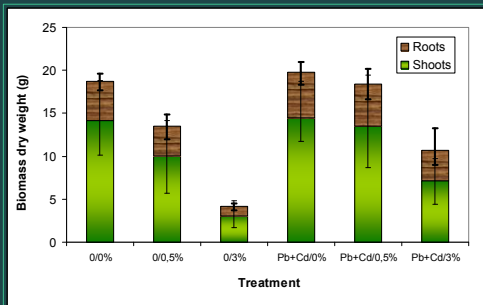
Cd Accumulation



Cd concentration in individual parts of *T. smyrnensis* at different soil salinities

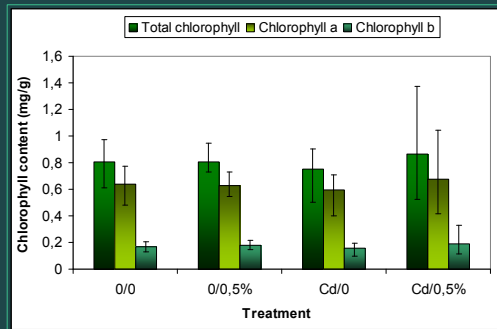
Salinity [%]	L/R
0	0.35
0.5	0.82
3	1.4

Tamarix smyrnensis

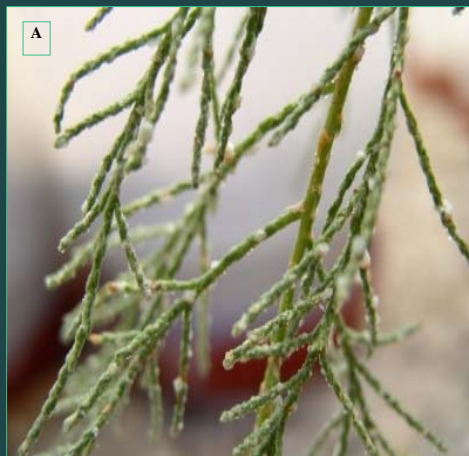


Biomass (dry weight)
Treatment with mixture of Pb & Cd at different salinities

Chlorophyll in the leaves
Treatment with Cd at different salinities

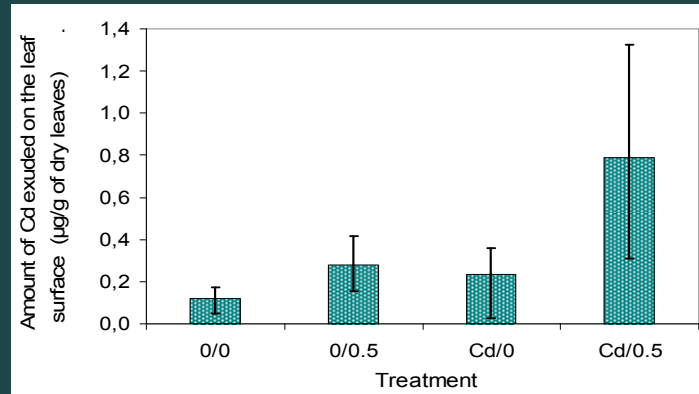


Tamarix: Salt crystals on leaves



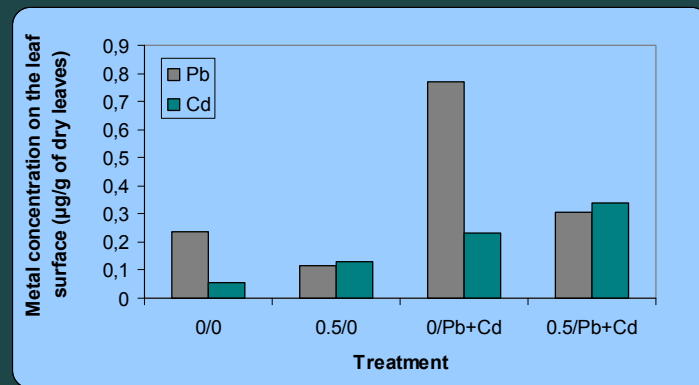
Droplets secreted by salt glands were crystallized on the leaves due to high Temperatures.

Cd Excretion by the Leaves



Cadmium excretion from leaf tissue of *T. smyrnensis* (pot experiment). Comparison of control plants and plant treated with 16 ppm Cd of dry weight of soil at two soil salinities (0% and 0.5%)

Pb & Cd Excretion by the Leaves



Metals excreted on the leaf surface of *T. smyrnensis* (pot experiment). Comparison of control plants and plant treated with 800 ppm Pb and 16 ppm Cd of dry weight of soil at two soil salinities (0% and 0.5%)

Heavy Metal Tolerance

Plant **mechanisms** of heavy metal tolerance:

- i. Avoidance
 - ii. Exclusion
 - iii. Immobilization
 - iv. Excretion
 - v. Mechanisms involving enzymatic changes
- ❖ The resistance of halophytes to salt stress is usually correlated with a more efficient antioxidant system (Zhu et al., 2004).
 - ❖ Thus, halophytes may be more capable to cope with heavy metals stress than common plants since heavy metal stress induces oxidative stress to cellular structures.

Excretion mechanism

- Salt secretion through salt glands is considered as an adaptive strategy to regulate plant tissue ion concentration
- An important mechanism which contributes to the resistance of all plants to increased salinity levels.
- **Halophytes** are adapted to saline environments:
 - salt avoidance
 - salt tolerance
 - salt evasion
- The main function of salt glands is the secretion of excess stress-inducing ions that invade the plant

Tamarix smyrnensis



- Species of the **genus *Tamarix*** are well known as **salt-tolerant** plants with the ability to **excrete excess salt as salt droplets through salt glands on their leaf surface**.
- There is evidence that the salt glands of *Tamarix sp.* secrete with **minimal selectivity** a variety of **different ions** and that the composition of the secreted salts is related to the composition in the rhizosphere.

Tamarix smyrnensis



Transverse section of the leaf of *T. smyrnensis* with salt gland

Salt crystals on leaf tissue of *T. smyrnensis* at different soil salinities



0% salinity



0.5% salinity

Hydroponic experiment

Experimental Conditions

Hydroponic growth with exposure to

100 ppm Pb and 5 ppm Cd

Age of plants: 10 months

Experimental period: 2 weeks

Temperature: 19 – 24°C

Humidity: 57 – 66%

Photoperiod: 12 h

Nutrient solution (mg/l):

143.0 $\text{Ca}(\text{NO}_3)_2$ 2.86 H_3BO_3

35.75 KNO_3 1.86 $\text{MnCl}_2 \cdot 4\text{H}_2\text{O}$

17.75 KCl 0.22 $\text{ZnSO}_4 \cdot 7\text{H}_2\text{O}$

35.75 KH_2PO_4 0.079 $\text{CuSO}_4 \cdot 5\text{H}_2\text{O}$

35.75 MgSO_4 0.6 $\text{FeSO}_4 \cdot 7\text{H}_2\text{O}$

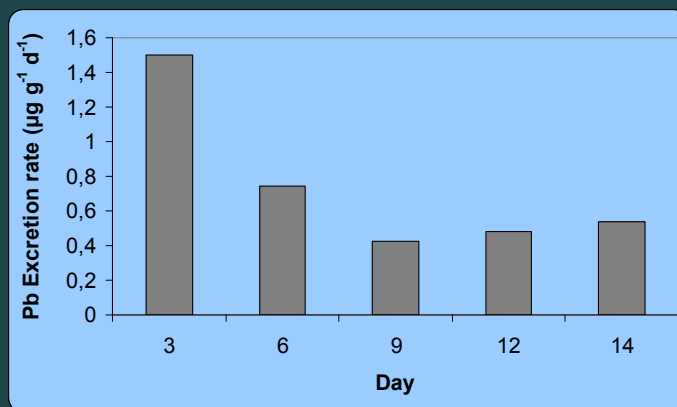


Hydroponic experiment

Measurements

- **Excretion rates of the metals** were measured by cleaning residues off leaf surfaces:
The area below the plant was covered by **weighted tissue paper**. In the **3rd, 6th, 9th, 12th and 14th day** the leaves were washed with 0.1% v/v **HNO₃** and the resulting solution was absorbed by the paper.
- **Metal content analysis in the paper wipes** was performed by ICP according to modified method of Soon
- **Metal content analysis in the plant tissue** was performed by ICP spectroscopy according to modified method of Soon
- **Determination of Pb and Cd content in the nutrient medium** was performed by ICP spectroscopy

Pb Excretion Rates



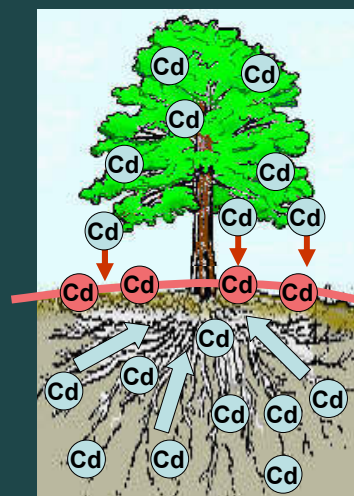
Pb excretion from leaf tissue of *T. smyrnensis* exposed to 100 ppm Pb and 5 ppm Cd (hydroponic experiment)

Phytoextraction of contaminated soils with heavy metals

- ❑ Problems of Phytoextraction
 - Contaminated crop disposal
 - Remediation time required
- ❑ Phytoexcretion process should be kept in mind
 - If not properly addressed, it reduces the effectiveness of other phytoremediation processes

Phytoextraction + Phytoexcretion

Opportunity to intervene (?)



Surface accumulation

Capture and remove on appropriate media

Phytoremediation processes:

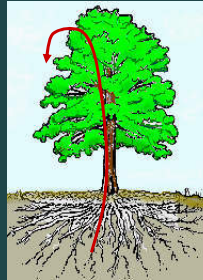
Phytoexcretion:

Excretion of heavy metals from the leaves

Phytovolatilization

Phytoextraction

Rhizofiltration



Phytodegradation

Phytostabilization

Enhanced Bioremediation (or Phytostimulation)

Phytoexcretion:

A Novel Approach of Phytoremediation (?)

□ "Phyto-Excretion":

- The plant can be viewed as a "biological pump" for heavy metals
- Intervening and capturing the droplets on suitable media before they are recycled onto the top soil

□ Advantages:

- ⇒ The frequency of tree pruning and uprooting is lowered
 - lower costs
 - faster remediation times
 - possibility of recovery of metals
- ⇒ Coupled to phytoextraction

Planning of Experimental part



Plant: *Nerium oleander*

Pot experiments
(10 weeks)

Cd (0, 0.5, 3% NaCl)

Pb (0, 0.5, 3% NaCl)

Pb & Cd (0, 0.5, 3% NaCl)

Pb increasing concentrations
(0, 0.5, 3% NaCl)

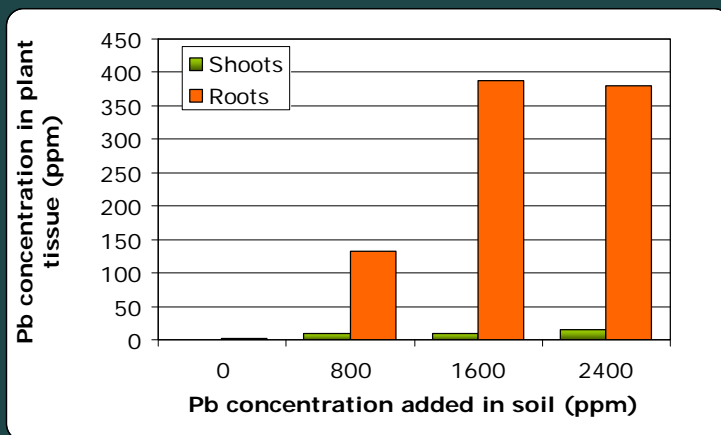
Hydroponic experiments
(2-4 weeks)

Pb increasing concentrations

Cd increasing concentrations

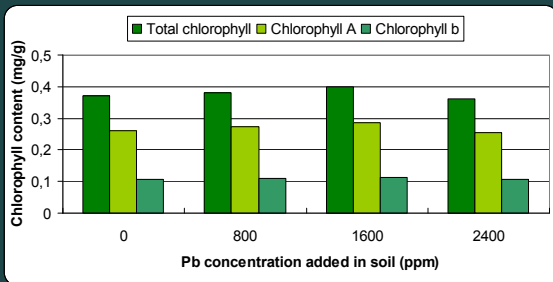
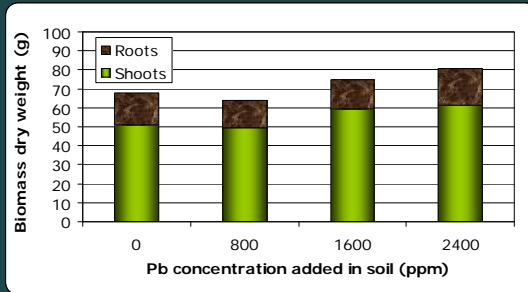
Nerium oleander

Pb concentration (mg kg^{-1} dry weight) in individual plants parts



Nerium oleander

Biomass (dry weight) of *Nerium oleander*

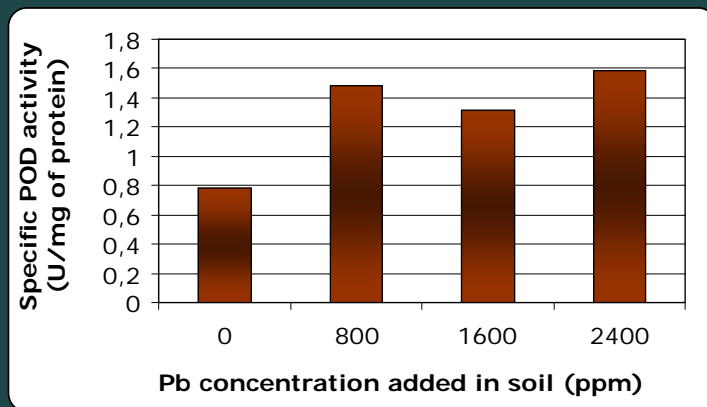


Chlorophyll contents of *Nerium oleander*

Nerium oleander

Is the plant under stress??

Effect of Pb on peroxidase activity of *Nerium oleander*

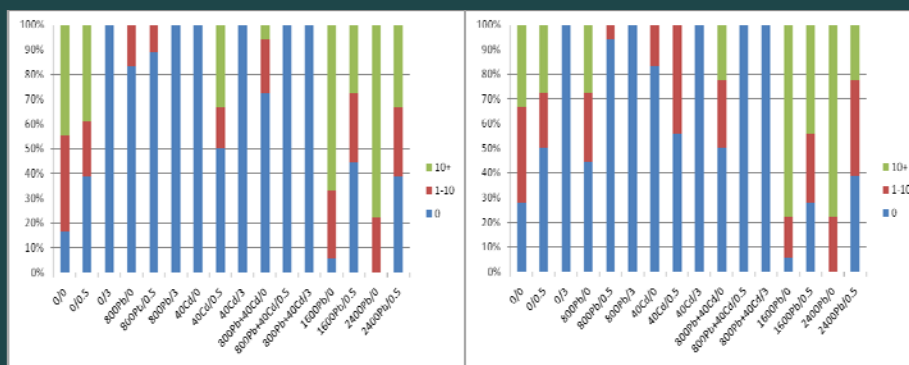


Interactions between Aphids (*Aphis nerii*) and Oleander Growing on Pb and Cd Contaminated Soil



Portion of plants *N. oleander* infested by aphids during weeks 5 to 7 and 8 to 10 are almost identical for all treatments.

(Plants infestation recording: no presence of aphids, plant infested by number of aphids from 1 – 10 and plant infested by >10 aphids)

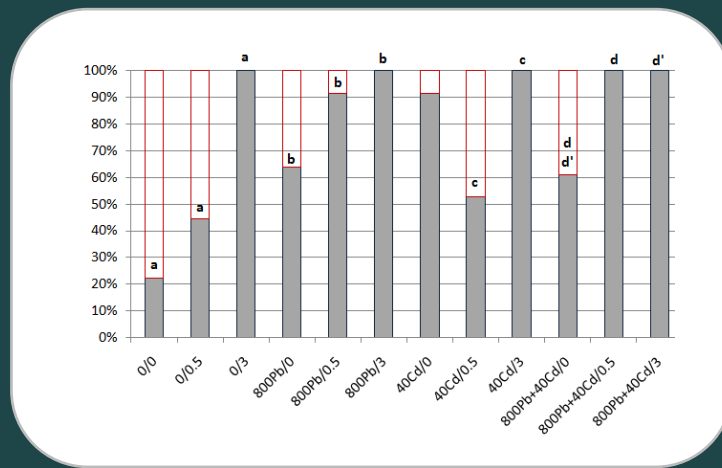


Weeks 5 to 7

Weeks 8 to 10

Portion of plants *N. oleander* not infested by aphids (P(X=0)) for various treatments with lead and cadmium as a function of salinity.

Portions marked with the same letter are significantly different with each other (corresponding to different saline concentrations) at least at 5% level of significance.



Overview of experimental results

Tamarix smyrnensis:

Suitable for phytoextraction in environments with increased salinity.

Nerium oleander:

A very good choice for phytostabilization.

Atriplex halimus:

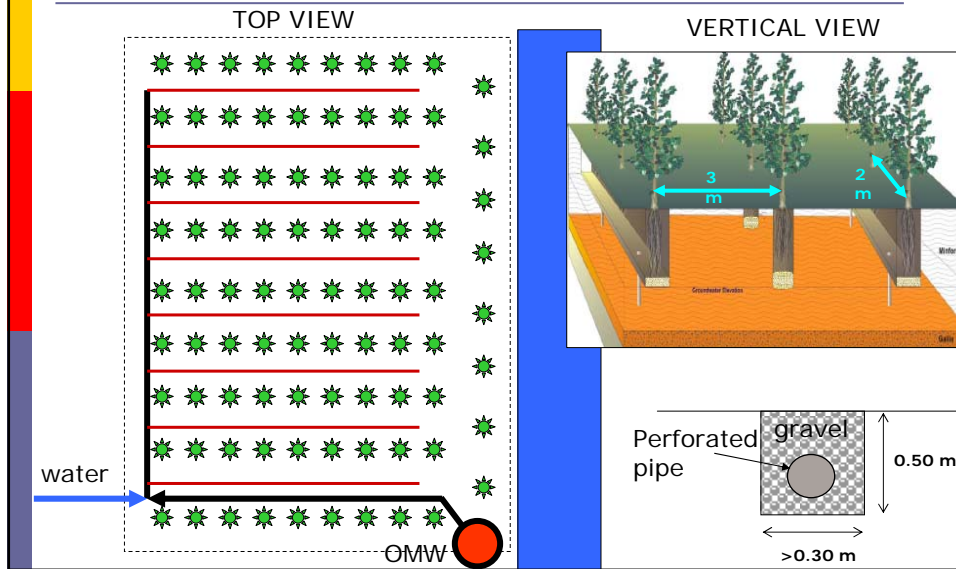
A new Pb-hyperaccumulator (?)

Kadukova, J., and N. Kalogerakis, "Lead accumulation from non-saline and saline environment by *Tamarix smyrnensis* Bunge", *European Journal of Soil Biology*, **43**, 216-223 (2007).

Manousaki, E., J. Kadukova, N. Papadantonakis and N. Kalogerakis, "Phytoextraction and Phytoexcretion of Cd by the Leaves of *Tamarix Smyrnensis* Growing on Contaminated Non Saline and Saline Soils", *Environmental Research*, **106**, 326-332 (2008).

Kadukova, J., E. Manousaki and N. Kalogerakis, "Pb and Cd Accumulation and Excretion by Salt Cedar (*Tamarix smyrnensis* Bunge)", *International Journal of Phytoremediation*, **10**, 31-46 (2008).

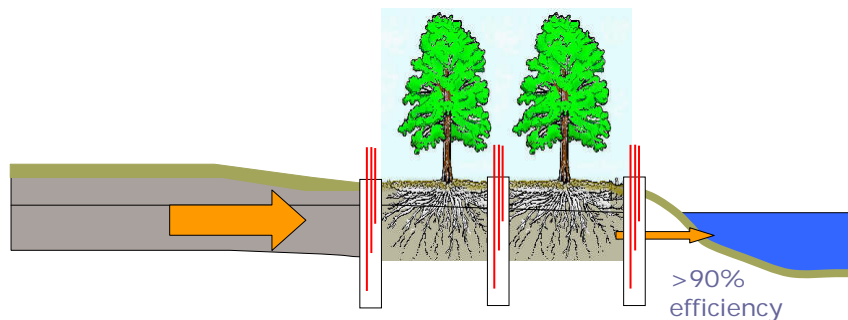
Phytoremediation of organics (OMW)





Protecting the river (riparian zone)

- Aim: To stop the pollutant plume and degrade contaminants that have been extracted by the plants
- Monitoring through multilevel wells



APPLICATION: Poplars to control the flow of nitrates from the agricultural land next to Evrotas river.

 ENVI Friendly

LIFE
Environment



Remediation of saline soils

- **Salinization** is one of the most serious problems confronting **sustainable agriculture in irrigated production lands** in semi-arid and arid regions. UN-EP estimates that ~20% of agricultural land and 50% of cropland in the world is salt stressed (Ravindran et al., 2007)
- Soils need proper amendments as a source of calcium (Ca^{2+}) to replace sodium (Na^+) from the cation exchange sites. The displaced Na^+ is leached from the root zone through excess irrigation (Qadir et al., 2003). **[Chemical remediation – Potential aquifer problems?]**
- Can phytoremediation help?

Phytoremediation of saline soils by halophytes

- Phytoremediation desalination approach #1
 - Cultivation of certain salt tolerant plant species with the **ability to increase the dissolution of soil calcite (CaCO_3) in the rhizosphere** to provide Ca^{2+} that can be exchanged with Na^+ at cation exchange sites. Displaced Na^+ can be leached out of the soil with irrigation water. (Qadir and Oster, 2002; Qadir et al., 2003; Qadir et al., 2004; Gerhardt et al., 2006) **[Aquifer problems?]**
- Phytoremediation desalination approach #2
 - Halophytes could be grown on salt-affected soils to **remove significant amounts of salt and Na^+ through their aerial parts**. Salt is removed from the soil to the extent that soil can be returned to agricultural productivity (Chaudhri et al., 1964; Gritsenko and Gritsenko, 1999; Owens, 2001; Keiffer and Ungar, 2002; Gerhardt et al., 2006; Ravindran et al., 2007).

CONCLUDING REMARKS

- ↪ There is a group of plants (halophytes) that have the capability to excrete heavy metals from their leaves as a detoxification mechanism.
- ↪ In this case, the plant becomes a “biological pump” for heavy metals. “Phyto-excretion” is an alternative phytoremediation process that should be further explored.
- ↪ The use of halophytes for phytoremediation applications should be further explored:
 - ❖ Rhizodegradation of organic contaminants [they can deal better with stress]
 - ❖ Rhizosphere enhanced bioremediation of mixed pollutants (metals + organics) [by removing the metals the microbes work better]
 - ❖ Soil desalination [a low cost long term remediation approach]



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